Physical & Chemical Changes Study Guide

Name: _____ Test Date: Friday January 27th

Vocabulary Term	Write the definition	Draw/insert a picture
matter		
mass		
particles		
substance		
gas		
solid		
liquid		
mixture		
physical change		
chemical change		
condense		
evaporate		
precipitate		

Answer the following True or False statements related to physical and chemical changes:

Ice melting to a liquid is a physical change
Water evaporating to a gas is a physical change
Mayonnaise being spread on bread is a chemical change
Rust forming on a bike is a chemical change
A firework exploding creates new heat and light, making it a physical change
Physical changes make new substances
Chemical changes can result in a change in odor or color
Physical changes can result in a change in shape or size
Ripping a piece of paper is a chemical change because it can't be reversed
A banana rotting is a physical change because it is still a banana

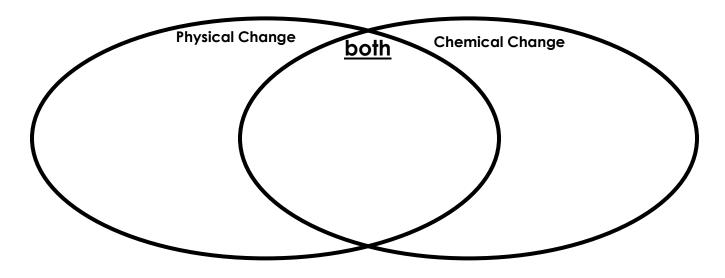
Draw an example below of two types of mixtures:

A heterogenous mixture and a solution (homogenous mixture)

Heterogenous Mixture	Homogenous Mixture (solution)	

Compare and Contrast Physical and Chemical Changes using the word bank and venn diagram below

Creates a new substance	No new substance created	<u>Mixture</u>	
Change in shape	Change in shape Change in color		
<u>Reversible</u>	<u>Nonreversible</u>	Change matter	



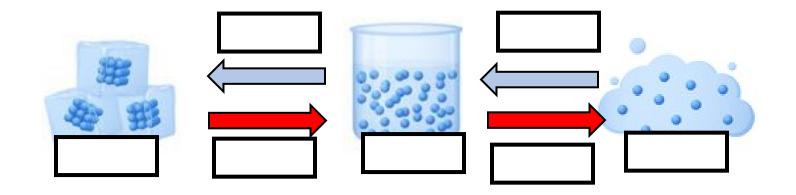
Show how the following items could go through a chemical change and physical change using the boxes below:

Object	Describe a chemical AND physical change it could go through and draw a picture of each		
	<u>Physical Change</u>	<u>Chemical Change</u>	
	Physical Change	Chemical Change	
	<u>Physical Change</u>	Chemical Change	

Fill in the following chart to show how water changes states of matter:

When water changes states of matter, it is showing a _____ change.

Word Bank				
o Solid (ice)	o Liquid	Gas (water	Evaporation	
	(water)	vapor)		
o Condensation	Melting	Freezing		



Locate all of the physical and chemical changes occurring in the following picture:

